

Team creates MRI for the nanoscale

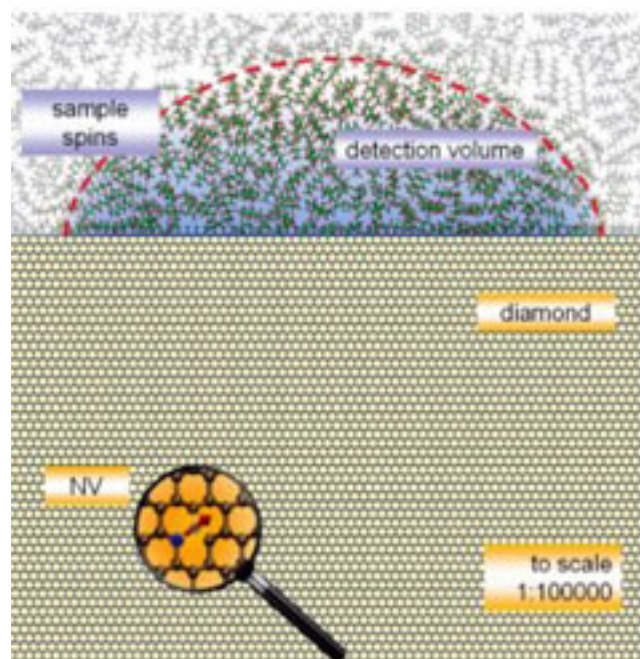


Image by T. Staudacher and F. Reinhard

Magnetic resonance imaging (MRI) reveals details of living tissues, diseased organs and tumors inside the body without x-rays or surgery. What if the same technology could peer down to the level of atoms? Doctors could make visual diagnoses of a person's molecules—examining damage on a strand of DNA, watching molecules misfold, or identifying a cancer cell by the proteins on its surface.

Now Dr. Carlos Meriles, associate professor of physics at The City College of New York, and an international team of researchers at the University of Stuttgart and elsewhere have opened the door for nanoscale MRI. They used tiny defects in diamonds to sense the magnetic resonance of molecules. They reported their results in the Feb. 1 issue of *Science*.

"It is bringing MRI to a level comparable to an atomic force microscope," said Meriles, referring to the device that traces the contours of atoms or tugs on a molecule to measure its strength. A nanoscale MRI could display how a molecule moves without touching it.

"Standard MRI typically gets to a resolution of 100 micrometers," about the width of a human hair, said Meriles. "With extraordinary effort," he said, "it can get down to about 10 micrometers"—the width of a couple of blood cells. Nanoscale MRI would have a resolution 1,000 to 10,000 times better.

To try to pick up magnetic resonance on such a small scale, the team took advantage of the spin of protons in an atom, a property usually used to investigate quantum computing. In particular, they used minute imperfections in diamonds.

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Diamonds are crystals made up almost entirely of carbon atoms. When a nitrogen atom lodges next to a spot where a carbon atom is missing, however, it creates a defect known as a nitrogen-vacancy (NV) center.

“These imperfections turn out to have a spin—like a little compass—and have some remarkable properties,” noted Meriles. In the last few years, researchers realized that these NV centers could serve as very sensitive sensors. They can pick up the magnetic resonance of nearby atoms in a cell, for example. But unlike the atoms in a cell, the NVs shine when a light is directed at them, signaling what their spin is. If you illuminate it with green light it flashes red back.

“It is a form of what is called optically detected magnetic resonance,” he said. Like a hiker flashing Morse code on a hillside, the sensor “sends back flashes to say it is alive and well.”

“The NV can also be thought of as an atomic magnet. You can manipulate the spin of that atomic magnet just like you do with MRI by applying a radio frequency or radio pulses,” Meriles explained. The NV responds. Shine a green light at it when the spin is pointing up and it will respond with brighter red light. A down spin gives a dimmer red light.

Professor Mireles has written on the theoretical underpinnings of the work and proposed the the project to the team, led by Professor Jörg Wrachtrup—a physicist at the University of Stuttgart in Germany—with the assistance of postdoctoral researcher Friedemann Reinhard and collaborators from the University of Bochum and the University of Science and Technology of China. Professor Wrachtrup heads a leading group studying such defects.

In the lab, graduate student Tobias Staudacher—the first author in this work—used NVs that had been created just below the diamond’s surface by bombarding it with nitrogen atoms. The team detected magnetic resonance within a film of organic material applied to the surface, just as one might examine a thin film of cells or tissue.

“Ultimately,” said Meriles, “One will use a nitrogen-vacancy mounted on the tip of an atomic force microscope—or an array of NVs distributed on the diamond surface – to allow a scanning view of a cell, for example, to probe nuclear spins with a resolution down to a nanometer or perhaps better.”

[Nuclear Magnetic Resonance Spectroscopy on a \(5-Nanometer\)³ Sample Volume](#) [1]

[Wrachtrup group](#) [2]

[2]

[Meriles group](#) [3]

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